PAPER • OPEN ACCESS

Fabrication of hexagonal photoreactor indoor lights

To cite this article: R Zainul et al 2019 J. Phys.: Conf. Ser. 1185 012007

View the article online for updates and enhancements.

You may also like

- Isotopic analysis of atmospheric hydrocarbons from CH4 to H2, CO and CO2 Elna J K Nilsson and M S Johnson

- Gas-phase optical fiber photocatalytic reactors for indoor air application: a preliminary study on performance indicators Ü Palmiste and H Voll

- Effect of natrium sulphate concentration on indoor lights photovoltaic performance R Yulis, R Zainul and M Mawardi

241st ECS Meeting

May 29 – June 2, 2022 Vancouver • BC • Canada Extended abstract submission deadline: Dec 17, 2021

Connect. Engage. Champion. Empower. Acclerate. Move science forward



This content was downloaded from IP address 103.216.87.250 on 07/12/2021 at 03:15

Fabrication of hexagonal photoreactor indoor lights

R Zainul^{1,*}, I Dewata² and B Oktavia³

¹Physical Chemistry Laboratory, FMIPA, Universitas Negeri Padang, Indonesia ²Environmental Chemistry Laboratory, FMIPA, Universitas Negeri Padang, Indonesia ³Analytical Chemistry Laboratory, FMIPA, Universitas Negeri Padang, Indonesia

*rahadianzmsiphd@yahoo.com

Abstract. Photoreactors are systems that convert light into an energy source. This research aims to design and build a visible light hexagonal photoreactor to generate electrical energy. The method used is geometry design and photoreactor prototype making. The tested photoreactors were tested using the integrated android F1-S (Appertude 2.0 Camera) light detector, and the resulting voltage and current measurements. The ability of hexagonal Photoreactor to convert visible light into electrical energy reaches a maximum condition of 16,728 mWatt in a light flux of 4573 lx. The light panel is characterized by FTIR, at 769.40 cm-1 wavelength, transmittance is only 23.39%, while at 921.34 cm-1, transmittance remains only 3.24%. In visible light, nearly 100% of the light can be passed on. Photoreactor hexagonal light space produces electrical energy reaches 0.0763835616 Watt/m².

1. Introduction

Photoreactor is a set of energy generating systems of light that are converted through a catalyst process[1,2]. In indoor light, the photoreactor is developed with an active nano-catalyst material at wavelengths between 400-600 nm. This effort can be done by modifying the material, such as copper into its oxide[3,4]. Another way is to dopan metal catalyst that reduces the band gap, so that the material works on visible light [5,6].

Photocatalyst applications not only on light conversion, on waste handling have been widely applied, especially in waste degradation [7,8]. Basically, the catalyst can be used to eliminate the presence of waste through transformation methods [7,8], for water splitting [9] and material modification ie PCC [10]. In contrast to heavy metal waste such as Pb, Cd and Cr, more effectively eliminated by adsorption method [11,12]. Some eliminant processes of this metal contaminant have also been carried out, such as the use of soil and organic extracts of unused parts, such as jengkol skin and other biomass.

In recent research, the application of catalysts to generate energy from visible light becomes interesting [13-16]. Researchers have made every effort to get photoreactors capable of converting room light into electrical energy such as design to materials, designing light-exposed panels, and assembling photoreactors in plenary [17]. In this research, the researcher wants to study the integrated design system in a hexagonal shaped frame, so that the light panel converter panel can be in one hexagonal place.

2. Experimental Section

2.1. Tools and Materials

Equipment used Sensor Light Detector with Camera Appertude (Brand Oppo), Volmeter/Ampere Meter Digital (Brand Heles), LED LV-45W 110-230 V, 50-60 Hz D17 410 mA /60 Lm/w (Hannoch), FTIR (PerkinElmer FTIR Spectrometer Frontier). The materials used are Na_2SO_4 (Merck), Aquadest, Copper Plate thickness 0.25mm, Alumunium Plate thickness 0.4mm, Glass with thickness 5 mm, Zinc Plate and Iron Frame 0.5cm diameter, and Silicon Glue.

2.2. Design of Photoreactor Hexagonal

The Photoreactor design is based on the size of the prepared electrode, which is 36.5 cm x 5 cm. The design is divided into a separate part of the wall, with the size of the container 8 cm x 40 cm as shown in Figure 1. The thickness of the panel is 15 mm with the contents of the panel 0.5 cm x 7 cm x 40 cm. The panel is assembled as shown in Figure 1.

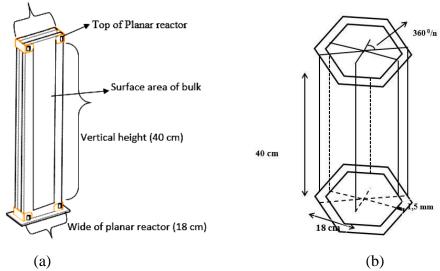


Figure 1. Planar Planet Sketch (a) and Hexagonal Photoreactor Scheme (b)

2.3. Assembling of hexagonal Photoreactor

The reactor base framework is a circle with a diameter of 36 cm, and 6 reactor sides have a side length of 18 cm. This is because on one side there are two photovoltaic panels. Thus there will be 12 photovoltaic panels to be arranged on a hexagonal prepared. After assembling the skeleton with desoldering, the hexagonal Photoreactor is paired as in figure 2.

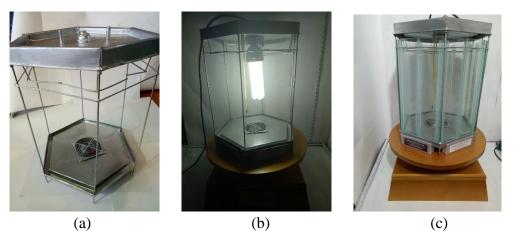


Figure 2. The photoreactor hexagonal framework (a) the source of the fluorescent lamp (b) and the photoreactor with the panel attached (c).

2.4. Photovoltaic Panel Setup

The panel is prepared with a copper oxide catalyst plate having a thickness of 0.25 mm as shown in Fig. 3. This preparation is carried out by thermal oxidation method at 400°C for 1 hour of combustion. Anode side used Alumunium plate with thickness of 0.40 mm, and copper oxide plate (Cu₂O) acts as cathode, while the source of electron comes from 0.5 M Na₂SO₄ solution in Gel form.

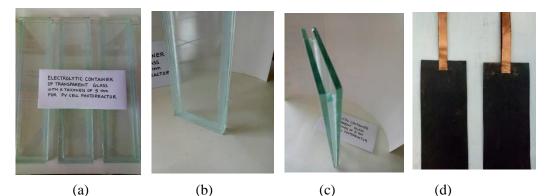


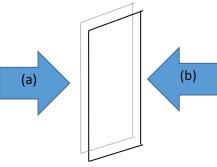
Figure 3. Front side reactor panel (a; b) side view (c) and copper oxide/Aluminum electrode (d)

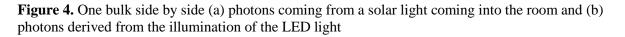
2.5. Preparation of Photovoltaic Panel

Bulk is the contents of a reactor made from Sodium Sulfate salt (Na₂SO₄) with a concentration of 0.5 M. This salt solution is formed into gel by using agar powder produced by PT. Agar Swallow. This flour is made from seaweed Gracilaria SP. The bulk making process is carried out by heating and stirring until it boils and then the solution is poured into the planar reactor and at the same time the CuO-Cu₂O/Al electrode plate is placed in a symmetrical position in the planar reactor housing. After the cooling is complete, the bulk will harden and the panel has been transformed into a photovoltaic panel that can generate electrical current.

3. Results and Discussion

The built hexagonal photonactor can produce a maximum voltage of 10.20 V using a series system on 12 hexagonal Photoreactor panels. The maximum current generated by series in 12 bulk units is 1.64 Ampere and maximum voltage of 10.20 mVolt, at an average flux of 4,573. On the inside there is a small gap with a distance of 3 mm to the outer and inner sides. The electrolyte will be inserted into the gap called the bulk or reactor. On one side of the Hexagonal there are two one-sided or two-sided bulk as shown in Figure 4. Since the hexagonal has 6 sides it requires 12 bulk units.





Photons derived from room light only activate wall a is caused on the inside (b) is the Aluminum plate. In part a, photons will cause the activation of electrons in the valence band to experience

excitation to the conduction band. In part a, a thin plate of CuO/Cu_2O will be used as an electron generator when exposed to photons from room light, so that the electrons in the valence band jump into the conduction band. The next process, in a closed circuit, the electrons in the conduction band will flow through the system in the hexagonal photoreactor. The conduction band's electron flow is what causes the photoreactor to have the ability to generate electrical energy.

The hexagonal photonactor constructed using 6 main wall blocks used as the side side of the Photoreactor. In principle, each side is a planar wall which is a unit of the planar reactor. In the manufacture of planar reactors, the effective surface area that activates the semiconductor is the surface of the CuO/Cu₂O plate, with a width of 5 cm and a height of 36.5 cm. Meanwhile, the bulk area that interacts directly with photons is 7 cm wide and 40 cm high. Based on this, one planar unit has an effective area of 0.01825 m², and a bulk interaction area of 0.0280 m².

Photoreactor hexagonal which has 6 sides has an effective surface area of 0.219 m2, ie 12 times 0.01825 m2. While the area of light room interaction in the hexagonal photoreactor bulk is 0.336 m2, ie 12 times 0.0280 m2. This broad calculation corresponds to the equation put forward by Bold. The equation is $A6 = \frac{1}{2} \times r^2 \times \sin \frac{360^\circ}{6}$ (Bold, B, 1969 : 61-62). The surface area is the total total surface of an object, which is calculated by summing the entire surface of the object.

Photoreactors are made to have a photoreactor finger with the calculation of cosine rules as follows:

Hexagonal radius of diameter =
$$\sqrt{\frac{2s^2}{1-\cos 60^\circ}}$$
 (1)

(The cosine rule)
$$s^2 = r^2 + r^2 - 2r^2 \cos \frac{360^\circ}{n} d = \sqrt{\frac{2(18 \text{ cm})^2}{0.5}}$$
 (2)

$$r = \sqrt{\frac{s^2}{2(1 - \cos 60^\circ)}} d = \sqrt{\frac{648 \text{ cm}^2}{0.5}}$$
(3)

Based on this calculation, the photoreactor finger is obtained 18 cm. This calculation is used to produce maximum photoreactor capability in converting light space. The material used as a panel wall is glass with the ability to forward the light of space close to 100%. The characterization of the glass material using FTIR as shown in Figure 5, where the glass material used (PT Asahimas production), has a 100% near transmitan capability in room light. At wavelength 769.40 cm⁻¹, transmittance is only 23.29%, while at 921.34 cm⁻¹ wavelength, transmitters are only 3.24% left. In this state, almost all light that interacts with the glass will be absorbed. Based on the FTIR characterization, the Photoreactor works well by using glass as a planar reactor panel.

IOP Conf. Series: Journal of Physics: Conf. Series 1185 (2019) 012007 doi:10.1088/1742-6596/1185/1/012007

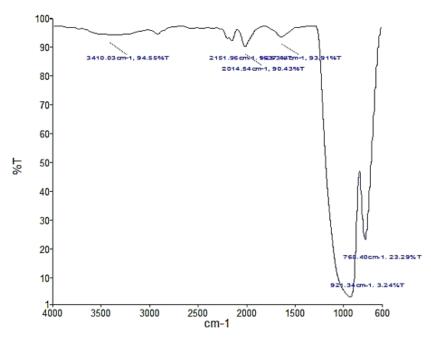


Figure 5. Result of characterization of glass material using FTIR

The ability of hexagonal photoreactors to produce maximum power is 0.0763835616 Watt/m². This energy is generated from 12 planar reactor units in a closed system in series, with a photon source at room light, at an average intensity of 4573 flux.

Photoreactors are made to have a surface area of electrons for photon reactions of 2160 cm², by calculation = (s - 8) (t - 4) n. The value of n is the number of the photoreactor side, and s is the width of the sides. Whereas, the surface area for the electrolyte reaction is = (s - 4) 40 n, so that the total reaches 3360 cm². The surface area of the reactor is = $\frac{1}{2}$ x r2 x sin $\frac{360^{\circ}}{n}$, and obtained for 35,073 cm². The hexagonal area (Hexagon) is = $\frac{n}{2}$ x r2 x sin $\frac{360^{\circ}}{n}$, ie 841.752 cm². The reactor volume is a hexagonal x height, ie $\frac{ns^2}{4}$ x $\frac{sin\frac{860^{\circ}}{n}}{1-cos\frac{860}{n}}$ x t, then obtained value 33670.08 cm³.

Figure 6. Hexagonal photocractors made with glass materials and copper oxide electrodes and gel electrolyte systems made from Na₂SO₄ salt solution.

4. Conclusion

Photoreactor Hexagonal made room light produces a maximum current of 1.64 mA and a maximum voltage of 10.24 V. Conversion energy production capability by utilizing light space reaches 0.0763835616 Watt/m². This capability can still be improved by improving the design and construction systems and materials used.

5. Acknowledgement

Thank you the Government of the Republic of Indonesia, Ministry of Technology, Research and Higher Education on the Research Grand of Riset Unggulan Perguruan Tinggi in 2018-2019. Special thanks to the Rector of the Universitas Negeri Padang, has been facilitated and funded for Terapan Research in 2018, and Chemical Laboratory of Universitas Negeri Padang. Thanks to the research team of UNP Chemistry undergraduate students and the Research Team from Zainul for Advanced Material Processing (ZAMP).

References

- [1] Zainul R, Alif A, Aziz H, Arief S, Dradjad S, Munaf E 2015 Design of Photovoltaic Cell with Copper Oxide Electrode by Using Indoor Lights, *Research Journal of Pharmaceutical*, *Biological and Chemical Sciences* 6(4) pp. 353-361
- [2] Bard A J 1982 Design of Semiconductor Photoelectrochemical Systems for Solar Energy Conversion, *The Journal of Physical Chemistry* 86 172-177
- [3] Chien-Lin Kuo R-C W, Jow-Lay Huang, Chuan-Pu Liu, Chun-Kai Wang, Sheng-Po Chang, Wen-Huei Chu, Chao-Hung Wang and Chia-Hao Tu 2009 The Synthesis and Electrical Characterization of Cu₂O/Al:ZnO Radial p–n Junction Nanowire Arrays, *Nanotechnology* (*IOP Publishing*) 20
- [4] Ping-Kuan Chang T-W K, Mau-Phon Houng, Chun-Hsiung Lu and Chih-Hung Yeh 2012 Effects of Temperature and Electrode Distance on Short-circuit Current in Amorphous Silicon Solar Cells, *IEEE* 978-1-4577-1829-8/12
- [5] Butler A W a K T 2014 Prediction of Electron Energies in Metal Oxides, *Account of Chemical Research* 47 364-372
- [6] Cook W H S a E L 1973 Compilation of Energy Band Gaps in Elemental and Binary Compound Semiconductors and Isolators, *J. Phys. Chem* 2 163-200
- [7] Zainul R 2016 Effect of Temperature and Particle Motion Against the Ability of ZnO Semiconductor Photocatalyst in Humic Acid *Der Pharmacia Lettre* 8 pp. 120-124
- [8] Zainul R 2016 Determination of the Half-Life and the Quantum Yield of ZnO Semiconductor Photocatalyst in Humic Acid *Der Pharmacia Lettre* 8 pp. 176-179
- [9] Zainul R, Alif A, Aziz H, Yasthopi A, Arief S, Syukri 2015 Photoelectrosplitting Water for Hydrogen Production Using Illumination of Indoor Lights, *Journal of Chemical and Pharmaceutical Research* 7(11) pp. 57-67
- [10] Mawardi M, Deyundha D, Zainul R 2018 Characterization of PCC Cement by Addition of Napa Soil from Subdistrict Sarilamak 50 Kota District as Alternative Additional Material for Semen Padang, *IOP Conference Series: Materials Science and Engineering* 335 012034
- [11] Anwar M, Munaf E, Kosela S, Wibowo W, Zainul R 2015 Study of Pb(II) Biosorption from Aqueous Solution Using Immobilized Spirogyra Subsalsa Biomass, *Journal of Chemical and Pharmaceutical Research* 7 715-722
- [12] Kurniawati D, Lestari I, Harmiwati S S, Chaidir Z, Munaf E, Zein R, Aziz H, Zainul R 2015 Biosorption of Pb (II) from Aqueous Solutions Using Column Method by Lengkeng (Euphoria logan lour) Seed and Shell, *Journal of Chemical and Pharmaceutical Research* 7 872-877
- [13] Madjene F, Aoudjit L, Igoud S, Lebik H, Boutra B 2013 A Review: Titanium Dioxide Photocatalysis for Water Treatment, *Transnational Journal of Science and Technology* 3
- [14] Navas J, Sa'Nchez-Coronilla A, Aguilar T, Hernández N C, Santos D M D L, Sánchez-Márquez J, Zorrilla D, Fernández-Lorenzo C, Alcántaraa R, Martin-Callejaa J 2014

IOP Conf. Series: Journal of Physics: Conf. Series 1185 (2019) 012007 doi:10.1088/1742-6596/1185/1/012007

Experimental and Theoretical Study of the Electronic Properties of Cu-doped Anatase TiO2, *Phys. Chem.* 16 3835-3845

- [15] Stepanov A L, Xiao X, Ren F 2013 Implantation Of Titanium Dioxide with Transition Metal Ions, *Nova Science Publishers, Inc.*
- [16] Xing Z, Zong X, Pan J, Wang L 2013 On the Engineering Part of Solar Hydrogen Production from Water Splitting: Photoreactor Design, *Chemical Engineering Science*
- [17] Zainul R 2016 Design and Modification of Copper Oxide Electrodes for Improving Conversion Coefficient Indoor Lights (PV-Cell) Photocells *Der Pharma Chemica* 8 pp. 388-395