



ICOMSET 2017

**2nd International Conference on
Mathematics, Science, Education
and Technology**

Padang, INDONESIA

October 5-6, 2017



PREFACE

On behalf of the Steering Committee, I would like to thank you for your participation in the 2nd International Conference on Mathematics, Science, Education and Engineering (ICOMSET2017) which has been held at Grand Inna Muara Hotel and Convention Center in Padang, West Sumatera, Indonesia from October 5 (Tuesday) through 6 (Friday), 2017.

This 2nd ICOMSET is organized by the Faculty of Mathematics and Natural Science, Universitas Negeri Padang. The main objective of this conference is to provide an international platform for researchers, Academicians as well as industrial professionals from all over the world to present Their research results in Mathematics, Science, Education, Technology, and other related fields. This conference also provides opportunities for the delegates to exchange new ideas and application experiences, to establish research relations and to find partners for future collaboration.

I would like to express my sincere appreciation to all the participants, financial sponsors, exhibitors, supporting organizations and all the committee members who has made ICOMSET 2017 successful. With these strong support, we are sure ICOMSET will be beneficial to all the participants, and you enjoy Padang.

We are looking forward to meeting you in the next ICOMSET.

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The Effect of Titanium Tetrahedral Coordination of Silica-Titania Catalyst on the Physical Properties of Biodiesel

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The Effect of Titanium Tetrahedral Coordination of Silica-Titania Catalyst on the Physical Properties of Biodiesel

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Abstract. This study investigates the correlation of the number of titanium tetrahedral coordination and biodiesel production. The solid-state method has been used to synthesis of silica-titania catalyst for biodiesel production, which the precursors, i.e. silica and titania commercials were heated in the temperature range of 450 - 550°C. The characterization of the prepared silica-titania has been studied by FTIR and DR UV-Vis in order to identify and calculate the presence of titanium tetrahedral coordination in silica-titania catalyst. A very small peak at around 950 cm⁻¹ indicated the presence of titanium tetrahedral coordination through Si-O-Ti bonds. Deconvolution of DR UV-Vis spectra showed the coordination of titanium in silica-titania is more octahedral. However, the number of titanium tetrahedral coordination of the prepared silica-titania is found higher than that of TiO₂ commercial. The increasing of titanium tetrahedral fraction in silica-titania affects the physical properties of biodiesel in terms of boiling point, viscosity and density, which is produced by the reaction of methanol and palm oil.

1. Introduction

The biodiesel production has developed on going with the reduction of natural resources from fossil fuel globally, high increase of world crude oil price and increasing environmental problem due to fossil fuel. Biodiesel is one of the most promising alternative fuels because it has several benefits over problems generated by fossil fuel. The added value of biodiesel is that its properties close to properties of diesel fuel; an engine using biodiesel does not need to be modified and lubricating property of biodiesel is better than that of diesel and therefore, it can overcome exhausted engine and maintenance cost of engine (1-3).

Palm oil is more promising source of biodiesel than other sources because its abundance in Indonesia is the most available in the world. The life time and production ability of palm trees are longer (until 25 years), and its resistance towards climate change as well. A report stated that the utilization of palm oil for biodiesel production not causing a reduction of palm oil available for consuming purpose (4,5).

The biodiesel production from transesterification of fatty acid and alcohol generally using a catalyst, either homogeneous or heterogeneous catalyst. Generally, the homogeneous catalyst is an acid catalyst (H₂SO₄) or base catalyst (NaOH and KOH). The utilization of homogeneous catalyst shows several drawbacks particularly in the separation process of catalyst and product. The product is



mixed with the catalyst used, thus, the separation process is not simple and requires high cost (not economics). When the separation is not good the biodiesel yield is still mixed with the catalyst and causes corrosion of engine (6). Several obstacles found in homogeneous catalyst can be overcome by the utilization of a heterogeneous catalyst. The biodiesel product has different phase from that of the catalyst and therefore, the separation process is more simple and economics. In addition, the used catalyst can be reused for next reactions (1).

Silica-titania ($\text{SiO}_2\text{-TiO}_2$) is one of binary oxides, which is important in the heterogeneous catalyst group. The utilization of silica-titania as a catalyst in the synthesis of biodiesel has not been widely reported. Based on the literature study, the silica-titania reported as catalyst in biodiesel fabrication was the substituted silica-titania with sulfate group. In general, the reaction product shows the higher catalytic activity of silica-titania and titania-sulfate with a conversion achievement of 90%. However, the percentage increasing of biodiesel yielded is not proportional with the percentage of sulfate group in silica-titania(7).

Several important matters such as the stability of silica-titania with sulfate group during calcinations, the correlation of sulfate composition and percentage of conversion, the existence of tetrahedral framework in silica-titania and stability of sulfate group during reaction of biodiesel fabrication have not been elucidated in previous work. The tetrahedral framework of titanium is known very active and determines the catalytic activity of titanium in silica-titania. The formation of tetrahedral framework through Si–O–Ti bond causes acidic properties on silica-titania surface increased due to different in geometrical form and coordination of Si and Ti (8). The acidic properties are much required for a catalyst in the fabrication of biodiesel.

This study reported the effects of titanium tetrahedral framework in silica-titania on the physical properties of biodiesel yielded. The synthesis of silica-titania is based on the solid state method because it is more eligible for solid precursors and low cost fabrication, compared to another method i.e. sol-gel method(9–11). Characterizations of samples applied FTIR and DR UV-Vis. The fraction of titanium tetrahedral in the synthesized silica-titania was calculated applying deconvolution of UV spectra and examined in biodiesel fabrication from the reaction of palm oil and methanol. The physical properties of biodiesel product including boiling point, viscosity, and density were determined.

2. Experimental

2.1. Chemicals

Titanium oxide (Acros), silica (Sigma Aldrich), and toluene (Merck) were purchased and used in this study. Palm oil (Bimoli) and methanol (Merck) were used for biodiesel synthesis.

2.2. Synthesis of silica – titania

Synthesis of silica-titania was done by mixing respective 10 mmol silica and 10 mmol titania (1 : 1) in 10 ml toluene. The mixture was sonicated using ultrasonic equipment for 1h to disperse silica and titania. Then the mixture was dried in a fume hood at room temperature to evaporate toluene. The solid sample formed was then calcinated at various temperatures (450°C, 500°C and 550°C) for 8h. The products from calcinations were characterized by FTIR and DR UV-Vis, and were applied for synthesis of biodiesel as well.

2.3. Characterization of silica-titania catalyst

The FTIR characterization was used to identify type of chemical bonding in the mixture of silica and titania. The samples were analyzed in wave number range of 4000 – 400 nm^{-1} . The DR UV-Vis characterization was used to determine tetrahedral and octahedral coordination in the sample and also to determine the fraction of titanium tetrahedral based on spectral deconvolution. The samples were analyzed in wavelength range of 200 – 400 nm.

2.4. Application of silica-titania catalyst for synthesis of biodiesel I(1,6,11-12)

In general, the synthesis of biodiesel was done through transesterification reaction of commercial palm oil and methanol with addition of silica-titania catalyst. The mole ratio of methanol and palm oil was made as 6 : 1, and the amount of catalyst was taken as 10% from the weight of palm oil. The mixture was homogenized by a magnetic stirrer and heated for 3h at 65°C. Then the mixture was cooled and the separation process was done between the product with catalyst and excess methanol. The first separation was done for catalyst by centrifuging. The next separation was done for excess methanol using rotary evaporator at temperature above boiling point of methanol. The biodiesel product was characterized with FTIR and the physical properties were examined for boiling point, viscosity, and density.

3. Result and Discussion

3.1 FTIR spectra of silica-titania prepared by solid state method

FTIR spectra of TiO₂, SiO₂ and series of SiO₂-TiO₂ synthesized based of various calcinations at 450°C, 500°C, and 550°C are shown in Figure.1. As seen in Figure.1, the spectrum of TiO₂ shows band absorptions at 734 cm⁻¹ and 2352 cm⁻¹. The band absorption at 2352 cm⁻¹ indicates the absorption of Ti-O, while the absorption at 734 cm⁻¹ is attributed to absorption of Ti-O-Ti. The spectrum of SiO₂ is attributed by the main band absorptions at wave numbers of 1058 cm⁻¹ and 803 cm⁻¹, respectively. The band absorption at wave number of 1058 cm⁻¹ is attributed to asymmetric stretching vibration of Si-O-Si, and that of 803 cm⁻¹ is related to symmetric stretching vibration of Si-O-Si (7)(14).

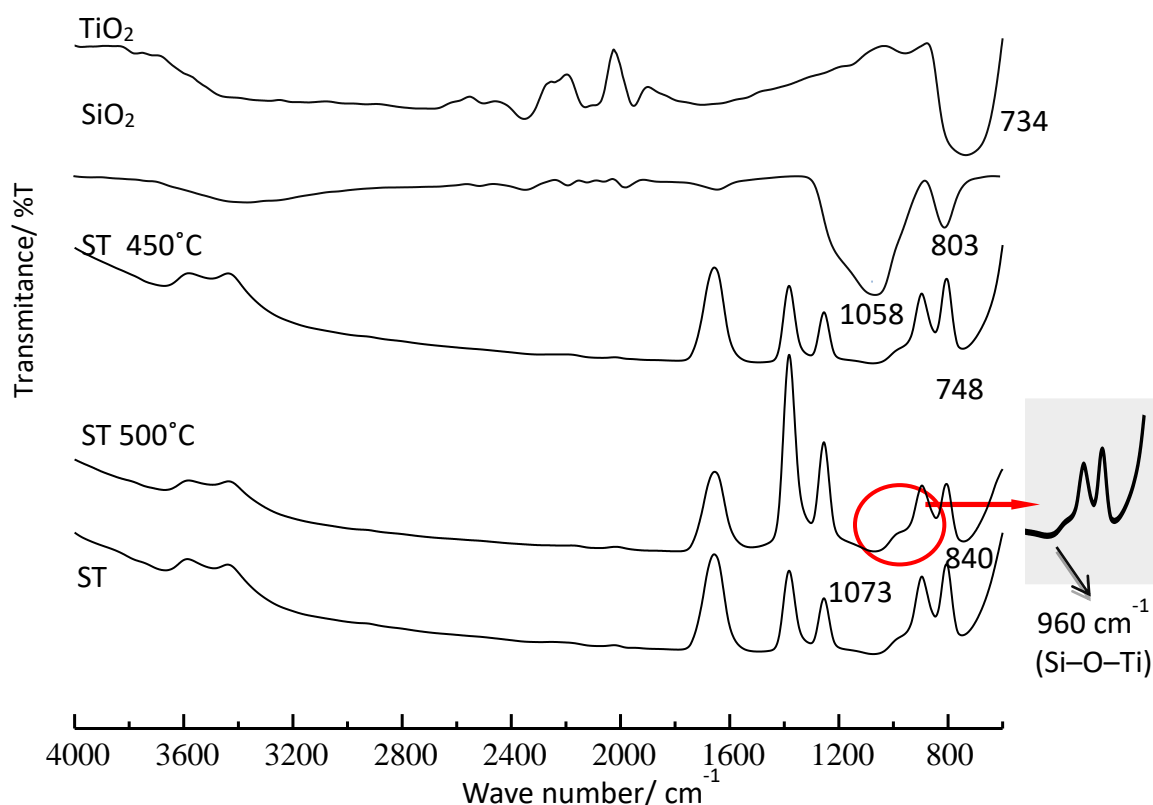


Figure 1. The FTIR spectra of TiO₂, SiO₂ and series silica-titania.

The FTIR spectra of SiO₂-TiO₂ series are presented by main band absorptions at respective wave numbers of 3668 cm⁻¹, 3429 cm⁻¹, 1073 cm⁻¹, 960 cm⁻¹, 840 cm⁻¹ and 748 cm⁻¹. The broad band absorptions at wave number of 3668 cm⁻¹ dan 3429 cm⁻¹ show the –OH stretching vibration of sample surface. The –OH band absorption is due to the existence of water vapor generated during FTIR characterization process (10). Other band absorptions observed are the weak broad band at wave number of 1073 cm⁻¹ and weak absorption band at 840 cm⁻¹. That band absorptions are related to asymmetric and symmetric stretching vibrations of Si-O-Si. In addition, a weak broad band absorption at wave number of 748 cm⁻¹ is attributed to vibration of Ti-O-Ti (7). A very small band absorption at wave number of 960 cm⁻¹ is due to vibration of Si–O–Ti bonding. This bonding is an indication of chemical interaction between SiO₂ and TiO₂. More bonding of Si–O–Ti formed indicated more titanium tetrahedral fractions generated in SiO₂ and TiO₂ (15). The formation of Si–O–Ti bonding is verified and confirmed by deconvolution band absorption of DR UV-Vis spectra (15–17).

3.2. Deconvolution of DR UV-Vis spectra

The deconvolution spectra of DR UV-Vis has been done to examine the effect of calcinations on titanium tetrahedral fraction and to evaluate the titanium tetrahedral fraction due to formation of Si–O–Ti bonding. According to literatures, the DR UV-Vis spectra in the range of 200 - <280 nm is attributed to tetrahedral region and that above than 280 nm is related to octahedral region as well as shown in Figure. 2 (10).

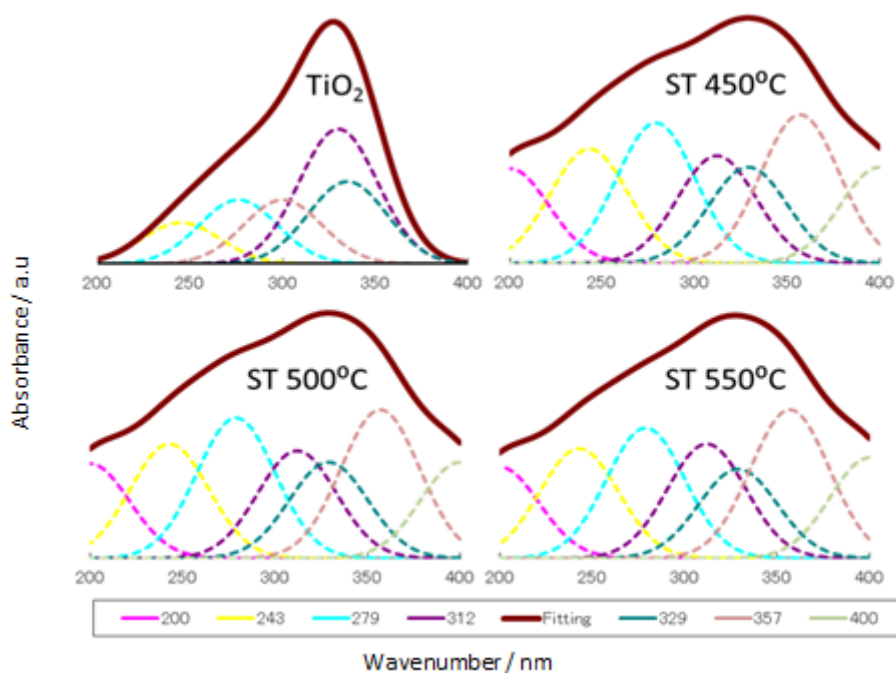


Figure 2. Deconvolution of DR UV-Vis spectra of TiO₂ and SiO₂-TiO₂ series

On the basis of deconvolution calculation the band absorptions at 200, 242 and 279 nm is corresponding to titanium tetrahedral fraction and that at 312, 329, 358 and 400 nm is related to titanium octahedral fraction. In general, silica-titania catalyst prepared by TiO₂ and SiO₂ solids is likely to form more octahedral fraction rather than tetrahedral fraction. However, the result of this investigation shows an increase of titanium tetrahedral fraction compared to TiO₂ commercial. The increasing of titanium tetrahedral fraction is shown in Table 1 (11) .

Table 1. Percentage of titanium tetrahedral based on deconvolution of DR UV-Vis spectra

Sample	Calcinations	Symbol	Tetrahedral
TiO ₂	-	TiO ₂	27,10 %
SiO ₂ -TiO ₂	450 °C	ST 450°C	43,03 %
SiO ₂ -TiO ₂	500°C	ST 500°C	42,11 %
SiO ₂ -TiO ₂	550°C	ST 550°C	41,87 %

On the basis of deconvolution DR UV-Vis spectra the titanium tetrahedral fraction increased from 27,10% to 43,03%. This phenomenon verified the formation of Si–O–Ti bonding due to interaction between SiO₂ and TiO₂, which was verified by FTIR spectra through the existence of band absorption at 960 cm⁻¹. Thus, it can be concluded that both results of FTIR and deconvolution of DR UV-Vis spectra are in good agreement.

3.3. Biodiesel synthesis

Generally, transesterification reaction produces biodiesel or methyl ester in liquid phase with glycerol as by-product (3). However, transesterification is not occurred but saponification with soap as main product. In order to confirm the formation of biodiesel at this preliminary investigation, several physical properties of the product are compared with that of its crude oil (Table 2). The characterization included density, viscosity and boiling point.

Table 2. Physical properties of crude oil and biodiesel product.

Sampel	Physical properties		
	Density (g/ml)	Viscosity (ml/s)	Boiling point (°C)
Palm Oil (Bimoli)	0,9055	0,0909	315
Biodiesel (ST 400°C)	0,8978	0,098	290
Biodiesel (ST 450°C)	0,8972	0,098	288
Biodiesel (ST 500°C)	0,8981	0,0961	290

Table 3 shows the results of density, viscosity, and boiling point. Based on the investigation, the biodiesel products have lower density and boiling point compared to that of palm oil. The result is in accordance with a literature that the target of this investigation is to reduce the physical properties of oil product through transesterification of palm oil. The synthesis process is to meet the purpose that the methyl ester produced by transesterification of palm oil can be advantaged as fuel in diesel engine without any change or modification of that engine. Figure. 3 shows the effect of titanium tetrahedral fraction on biodiesel properties

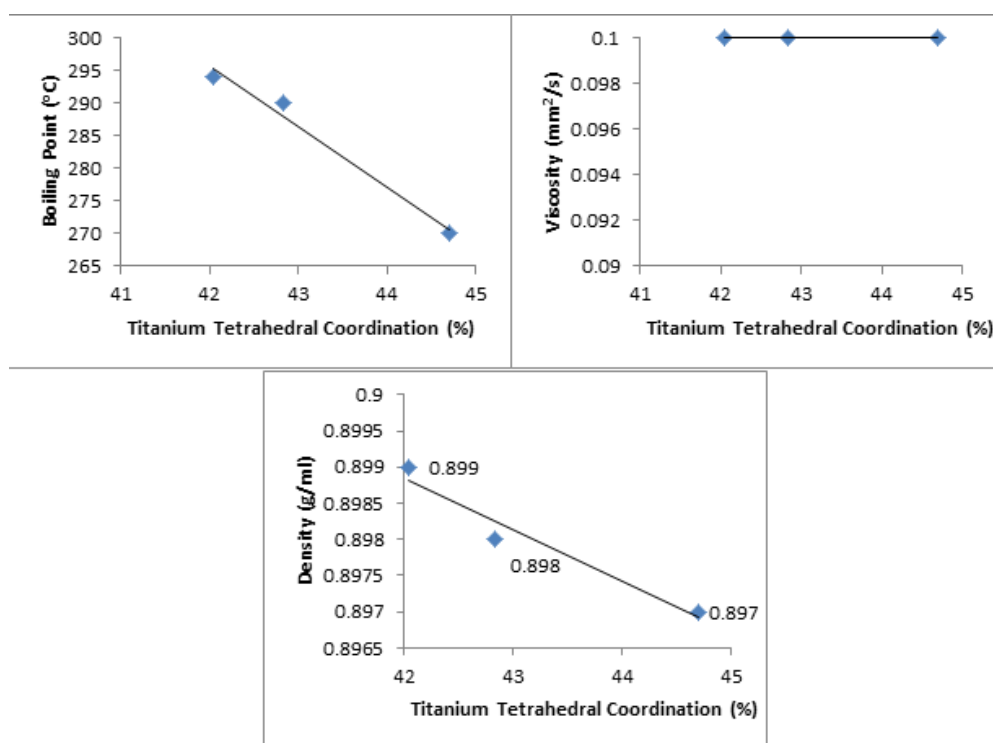


Figure 3. The correlation between titanium tetrahedral fraction in catalyst SiO₂-TiO₂ with physical properties of biodiesel.

Figure. 3 shows the increased titanium tetrahedral fraction in silica-titania catalyst causing decreased in both density and boiling point of biodiesel product but increased in its viscosity. This investigation is attributed to higher the quantity of titanium tetrahedral fraction causing better catalytic activity of silica-titania catalyst. The obtained results is in accordance with previous studies described that catalytic properties of silica-titania is influenced by its local structure, i.e. the tetrahedral as active local structure (10).

In addition, the FTIR spectra is also used to examine the change of functional groups from crude oil to methyl ester as shown in Figure.4 with respect to biodiesel product (450°C). The substantial absorption peaks of palm oil (bimoli) are appeared at wave number of 2922.15 cm⁻¹, 2859.99 cm⁻¹, 1742.80 cm⁻¹, 1453.60 cm⁻¹ and 1160.60 cm⁻¹ and the remarkable absorption peaks of biodiesel product (450°C) are attributed at wave number of 2922.02 cm⁻¹, 2859.89 cm⁻¹, 1742.45 cm⁻¹, 1453.49 cm⁻¹ and 1160.54 cm⁻¹. The strong band absorption at wave number 1800 – 1700 cm⁻¹ is attributed to C=O, while the band absorption at wave number 1170 – 1200 cm⁻¹ is related to C-O-C vibration of methyl ester and weak absorption at 1600 – 1450 cm⁻¹ is corresponding with vibration of C-C group. In addition, strong band absorption at 3000 – 2800 cm⁻¹ is related to C-H vibration of methyl group (18).

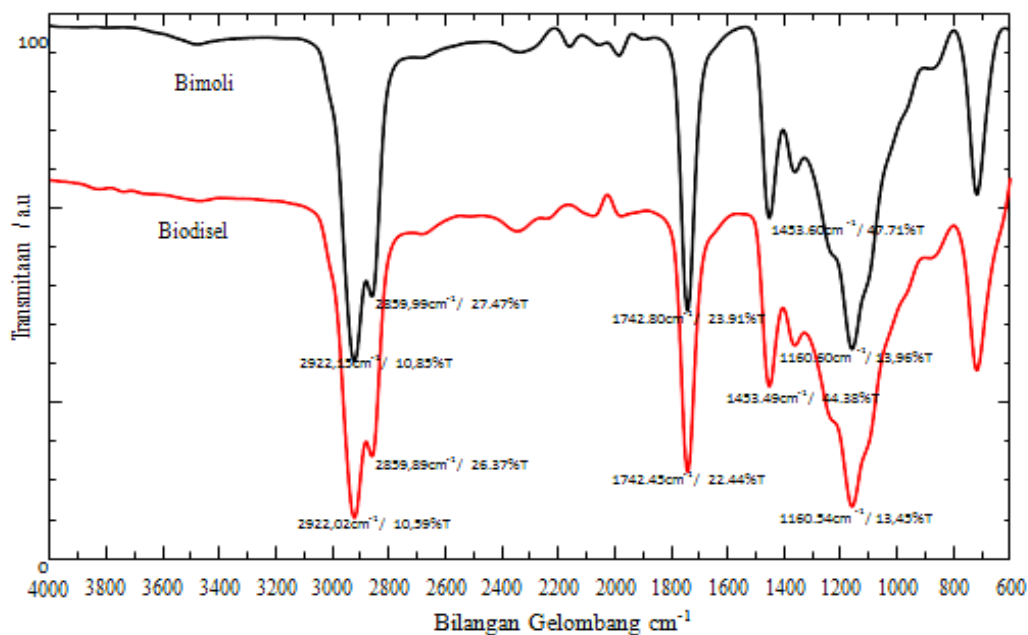


Figure 4. FTIR spectra of crude oil and biodiesel

4. Conclusion

The synthesis of silica-titania from solid precursors using solid-state method can increasing the number of titania tetrahedral framework in silica-titania, compared to the commercial titania. The highest number of titanium tetrahedral obtained by calcining the mixture of titania and silica precursors at 450°C for 8hr. The properties of biodiesel formed during transesterification between palm oil and methanol affected by silica-titania. It was proved by the determination of boiling point, viscosity and density.

Acknowledgement

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